

Cursos De Carpinteria Gratis Para Descargar

N Colangelo

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represent the reactants and products in
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equations where it shows the different number of elements is conserved in a reaction in a nuclear reaction the atomic mass and proton number are conserved

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web a nucleus changes into a new element by emitting nuclear radiations these changes are described using nuclear equations example 86 219 $\text{textrm{rn}} \rightarrow \text{textrm{po}}$ 84 215

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it s radioactive it s going to undergo spontaneous decay it s going to try to get a better neutron to proton ratio so let s look at the nuclear equation which represents the spontaneous decay of carbon 14 so here is

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 $\text{li s} + \text{n}_2 \text{ g} \rightarrow \text{li}_3 \text{ n s}$
 $\text{ca oh}_2 + \text{hbr} \rightarrow \text{cabr}_2 + \text{h}_2 \text{ o}$
 $\text{c}_4 \text{ h}_{10} \text{ o} + \text{co}_2 \rightarrow \text{h}_2 \text{ o} + \text{nh}_3$
 $\text{cuo} + \text{cu} \rightarrow \text{n}_2 \text{ h}_2 \text{ o}$
 $\text{q}_{10} + \text{co}_2 \rightarrow \text{h}_2 \text{ o} + \text{as}$
 $\text{naoh} + \text{na}_3 \text{ aso}_3 \rightarrow \text{h}_2$

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web sep 12 2022 the half life of the nuclei and thus the material is $t_{1/2} = 693$ million years $\lambda = 1.99 \times 10^{-5} \text{ s}^{-1}$

1 b since strontium 91 has an atomic mass of 90.90 g

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