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a write the balanced chemical equation for the reaction that
occurs when solutions of barium chloride and sodium
carbonate are mixed refer to table 1 on page 437 in chapter
13 for solubility $\text{BaCl}_2 \text{ aq} + \text{Na}_2\text{CO}_3 \text{ aq} \rightarrow \text{BaCO}_3 \text{ s} + 2\text{NaCl aq}$ b to
which of the five basic types of reactions does this reaction
belong double displacement 10

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section 2 chemical equations a chemical equation uses
symbols to represent a chemical reaction and shows the
relationship between the reactants and the products of a
reaction in words carbon dioxide and water react to form
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