

Diferencia Entre Numero De Oxidacion Y Valencia

Guido Guayasamin Guerrero

Química analítica cualitativa (reacciones en solución) Vicente Bolaños Chombo, 2003

Química i ,

Química básica Alejandrina Gallego Picó, 2018-10-08 El texto, básico pero con rigor, ofrece una exposición de los principios de la Química, accesible incluso para estudiantes sin apenas conocimientos de esta materia. A su parte teórica se une una completa y amplia colección de ejercicios, algunos explicados detalladamente y resueltos; y, otros, propuestos e indicada su resolución, Los autores de este libro son profesores de la UNED y han elaborado diversos materiales didácticos adaptados a la enseñanza a distancia de materias regladas y no regladas.

Tratado de química orgánica Federico Klages, 2006

Operaciones básicas de laboratorio CASADO SANCHEZ, EVA MARIA, DURAN BARQUERO, PURIFICACION, MIRO ARIAS, TRINIDAD, PAREDES DE LA SAL, ANTONIO JOSE, 2012-07-01 En este libro se desarrollan los contenidos curriculares correspondientes al módulo profesional de Operaciones Básicas de Laboratorio, dirigido a los alumnos que cursan el Ciclo Formativo de grado medio de Farmacia y Parafarmacia, perteneciente a la familia profesional de Sanidad.; Los contenidos que se imparten en este módulo son muy procedimentales de manera que aportarán al alumno la formación necesaria para realizar su actividad profesional en un laboratorio del ámbito farmacéutico, así como unos conocimientos y habilidades fundamentales que le pueden ser de gran ayuda en estudios superiores incluidos en la rama sanitaria y científica.; Con este texto se pretende, además, construir una base de conocimientos y destrezas para el posterior desarrollo del módulo de Formulación Magistral.; Para la elaboración de este texto se han tomado como base los contenidos establecidos en la legislación vigente, tanto nacional como autonómica, estableciendo como pilares fundamentales los siguientes puntos;.- Preparación y mantenimiento de materiales y equipos de laboratorio.;- Preparación de disoluciones.;- Separación de sustancias.;- Identificación y caracterización de sustancias.;- Toma de muestras y estudio de las mismas.; Además, en cada unidad didáctica se han diseñado diferentes recursos y actividades de enseñanza-aprendizaje como: experiencias de laboratorio, actividades propuestas y resueltas, autoevaluaciones, enlaces de internet, cuadros de recuadra y sabías que, de gran utilidad para afianzar los conocimientos teóricos.; Por último, al final del texto se proponen una serie de prácticas de laboratorio para que el alumnado relacione los conocimientos teóricos con la realidad, que es el trabajo en el laboratorio

farmacéutico.

Química 1. Fundamentos ,

Química, un proyecto de la A.C.S. American Chemical Society,2005 Este texto de Química está desarrollado y escrito por un grupo de excelentes educadores químicos escogidos por la American Chemical Society. Poniendo un constante énfasis en el aprendizaje activo, la obra propone a los estudiantes que no lean simplemente sobre el mundo de la química, sino que lo exploren de manera personal. Química es un libro esencial para aquellos profesores que quieran incorporar a su docencia el aprendizaje basado en la investigación.

Química y física en la informática Martínez Riachi Susana H.,2023-03-17 Esta obra trata sobre cómo la evolución de los descubrimientos y los conocimientos referidos a algunos aspectos de química y física han sido los pilares fundamentales para el desarrollo de tecnologías que hicieron posible disponer hoy de dispositivos y sistemas de información cada vez más sofisticados. Es un libro diseñado específicamente para ser ...

Química Rosa María González Muradás, Pilar Montagut Bosque, Maria Del Carmen Sansón Ortega, Roberto René Salcedo Pintos, 2014-10-21 Al inicio de cada unidad se presenta una breve introducción histórica con los fundamentos teóricos que requiere el estudiante para desarrollar el tema, mismos que aplicará al resolver los diversos problemas que se establecen. Contiene referencias que amplían y profundizan el tema que se cubre. En seguida se incluyen problemas resueltos que se comprenden mejor con las explicaciones de las estrategias de resolución. Después, planteamos problemas complementarios con sus respuestas al final del texto, a los que se incorporan la sección Alerta que se intercala a lo largo del problema, cuando lo consideramos necesario, para evitar errores comunes. Los problemas están ordenados en una secuencia ascendente en cuanto a su grado de complejidad. Al final del texto incorporamos referencias bibliográficas, hemerográficas y electrónicas; cuya consulta favorecerá a profundizar en los conocimientos que el estudiante considere necesarios.

Tabla periódica en espiral y propiedades zonales L. A. Bravo, 2025-05-12 El autor, Prof. Dr. Luis Bravo, con una gran experiencia pedagógica tanto en su país como en diversas Escuelas y Universidades latinoamericanas y europeas comenzó esta labor en el año 1947 sobre un trabajo publicado en el Journal of Chemical Education por Irving Gordon que sirvió de base para trabajar de modo sistemático y muy claro en la enseñanza de la Química general e inorgánica a todos los niveles (Enseñanza Media. Preparatoria y Facultad). Los alumnos colaboraron reproduciendo las tablas en murales efectuando una labor de equipo. Los cuadros del mencionado autor tratan los temas de valencia, actividad de los metales, grupos analíticos, densidades, puntos de fusión y ebullición.

Química inorgánica moderna María del Consuelo Alcantara Barbosa, 1972

Química. La ciencia básica DOMINGUEZ REBOIRAS, MIGUEL ANGEL, 2006-01-01 Esta obra se ha escrito pensando en las necesidades de los estudiantes de los primeros cursos de Facultades de Ciencias, Farmacia, Ciencias Ambientales y de la

Salud, y Escuelas Tecnicas.

Química. Fundamentos experimentales. Guía del profesor Robert W. Parry,1974 Esta Guía está destinada a proporcionar apoyo al profesor y a establecer claramente los principios y las razones que guiaron a los autores en la programación y desarrollo del libro Química, fundamentos experimentales.

Tratado de química inorgánica, fundamental y sistemática Modesto Bargalló,1962

Introducción al estudio de la corrosión y la protección de metales Damiana I Martín Laurencio ,María Pacheco Gola ,2024-12-29 Introducción al estudio de la corrosión y protección de metales, en su segunda edición, responde al propósito de disponer de un texto actualizado, ajustado al programa de la asignatura Corrosión y protección de metales, del nuevo plan de estudios para la carrera de Ingeniería Química, iniciado en el curso académico 2019-2020. En sus 12 capítulos se sigue una estructura consecuente con la metodología recomendada para el estudio de los problemas de corrosión, enriquecida con los criterios y experiencias de esta labor en la industria cubana, que hacen de este texto un material imprescindible en la formación técnica y profesional. La organización lógica y coherente, así como una razonada presentación de todos los aspectos científicos y técnicos de los contenidos tratados hacen de esta una obra de mucho valor, lo que se complementa con ilustraciones, preguntas y problemas que refuerzan su carácter didáctico y práctico.

Principios de Química. La búsqueda del conocimiento. P. W. Atkins,Loretta Jones,2006

PRINCIPALES COMPUESTOS QUÍMICOS CLARAMUNT VALLESPÍ Rosa M^a ,CORNAGO RAMÍREZ Pilar ,ESTEBAN SANTOS Soledad ,FARRÁN MORALES Angeles ,PÉREZ TORRALBA Marta ,SANZ DEL CASTILLO Dionisia ,2015-07-07 La asignatura Principales Compuestos Químicos se incluye dentro del módulo de Formación Básica del Grado en Química. El objetivo general de dicha asignatura es proporcionar un conocimiento básico, por una parte, de la reactividad y propiedades de los compuestos orgánicos y biomoléculas y, por otra, de la química de los elementos de los grupos de la Tabla Periódica y compuestos más importantes a los que dan lugar, así como de las reacciones propias de las sustancias radiactivas. También en formato e-book.

Química: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear

Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Principios básicos de química Harry B. Gray, Gilbert P. Jr. Haight, 2021-07-01 Este texto es una respuesta al reto que significa presentar una relación innovadora y moderna de los principios de la Química. El rápido desarrollo de los últimos cursos preuniversitarios está creando presiones sin precedentes que obligan a cambiar la enseñanza en el curso de iniciación. En este nivel se necesitan urgentemente aproximaciones al estudio de los campos más avanzados de la Química, con una presentación puesta al día. Además del texto propiamente dicho, existe la Guía para el profesor.

Química, una versión moderna Guido Guayasamin Guerrero, 1979

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