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web dec 8 2019 charles law is a special case of the ideal gas law in which the pressure of a gas is constant charles law states that volume is proportional to the absolute temperature of a gas at constant pressure doubling the temperature of gas doubles its volume so long as the pressure and quantity of the gas are unchanged

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web solution v1 2 50 l v2 t1 273 30 303 k t2 273 70 343 k we will be applying charles law as pressure is constant here v1 t1 v2 t2 or v2 v1 t1 x t2 2 50 l

 $303\ k\ x\ 343\ k\ 2\ 83\ l\ at\ 90\ c\ a\ helium$ sample has a volume of 500 ml

determine the temperature at which $% \left(\mathbf{r}\right) =\left(\mathbf{r}\right)$

the volume of the liquid will become $240\ ml$