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measure an appropriate volume of each liquid eg 25 cm<sup>3</sup> place one of the *enthalpy change of neutralization chemistry libretexts* - Apr 29 2023 web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol<sup>-1</sup> *acids and alkalis aqa synergy required practical bbc* - Jul 01 2023

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 hcl aq + naoh aq → nacl aq + h<sub>2</sub>o l for which ΔH = 57 kJ mol<sup>-1</sup> eg 2 h<sub>2</sub>so<sub>4</sub> aq + 2 koh aq → k<sub>2</sub>so<sub>4</sub> aq + 2 h<sub>2</sub>o l for which ΔH = 57 kJ mol<sup>-1</sup>

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