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web apr 23 2008 1 3 8 the temperature setting on forced air warming devices should be set at maximum and then adjusted to maintain a patient temperature of at least 36 5 c 2008 1 3 9 all irrigation fluids used intraoperatively should be warmed in a thermostatically controlled cabinet to a temperature of 38 c to 40 c 2008

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symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

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equations our goal is to have the same number of each type of atom on both sides of the equation only change the coefficients these are the numbers in front substances never change the subscripts the small numbers after elements

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magnizium in elemental form h 2 there are 2 molecules of nitric acid each acid has one hydrogen so two molecules of nitric acid is 2 hydrogens n 2 there are 2 molecules of nitric acid each acid has one nitrogen so two molecules of nitric acid is 2 nitrogens o 6

solved 1 enter balanced molecular equation for the reaction

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including phases 2 enter a net ionic equation for the reaction between nitric acid and magnesium hydroxide express your answer as a chemical equation including

mgco3 hno3 mg no3 2 h2o co2

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web magnesium carbonate nitric acid magnesium nitrate water carbon dioxide one mole of magnesium carbonate mgco 3 and two moles of nitric acid hno 3 react to form one mole of magnesium nitrate mg no 3 2 one mole of water h 2 o and one mole of carbon dioxide co 2 show chemical structure image reactants

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web magnesium is already balanced on the reactant and product side balancing nitrogen by multiplying hno 3 by 2 we get mg s magnesium 2 hno 3 l nitric acid mg no 3 2 s magnesium nitrate h 2 g hydrogen gas oxygen and hydrogen get automatically balanced by balancing nitrogen

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 2 no2 h2o and provide the correct
 coefficients for each compound to
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 balanced total equation mg oh 2 aq 2
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 ionic equation mg 2 2 oh 1 2 h 1 2no 3
 1 2h 2 o mg 2 2 no 3 1 net ionic
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