

# Regolamento Gioco Scala 4

**R Barnett**

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1805 was the prince regent and the de  
facto ruler of the kingdom of kottayam  
in malabar india between 1774 and  
1805 he led the pychy rebellion wynaad  
insurrection coiote war against the  
english east india company he is  
popularly known as kerala simham lion

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low budget commercial movie but not  
an arthouse film in its original usage  
during the golden age of hollywood the  
term more precisely identified films  
intended for distribution as the less  
publicized bottom half of a double  
feature akin to b sides for recorded  
music

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thrissur folksong performance of  
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india the music of kerala has a long and  
rich history it is not the same as  
malayalam poetry although most of it is  
poetry driven kerala has a rich tradition  
in carnatic music songs formed a major  
part of early malayalam literature  
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history dates back centuries and its  
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indicates that when the pressure of the object increases the volume of the object decreases and vice versa because volume and pressure relationship is inversely proportional while the temperature is constant  
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 web overall the experiment succeeded in showing that temperature and pressure for an ideal gas at constant volume and mass follow the relation of the ideal gas equation differences existed in the experimental graph of temperature versus and pressure and the theoretical curve of temperature versus pressure  
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 web we then changed the volume between 5ml and 20ml and recorded

the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure  
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 web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor  
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 web he concluded that at a constant

pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c  
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 summary the purpose of this experiment is to be able to understand the relationship between the pressure volume and temperature of a gas understanding this relationship is important as it allows  
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 web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature

into the ideal gas equation and solve for  $r$

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant  $k_p$  t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid  
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determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts