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away from the iron or steel oxygen can be excluded by storing the metal in an atmosphere of unreactive nitrogen or argon a desiccant is a

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there are the following commonly used methods of
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galvanizing chrome plating cladding silver plating gold
plating nickel plating copper plating rhodium plating

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electrochemical corrosion reactions on a metal surface
mechanism corrosion inhibitors work by one or more of the
following mechanisms they adsorb on metal surfaces to form
protective films they combine with corrosion product films to
protect metal surfaces

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attack corrosion is usually to put a protective barrier
between the metal and the corroding agents this can be a
wide variety of things paint an oil sealant or an
electrochemical solution like a galvanized zinc coating

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prevent corrosion and they shall know how to prevent
corrosion the method of detailing the structural element
shall be done in a manner that it does not encourage
corrosion the following key aspect can be highlighted
adequate access

corrosion prevention and protection methods

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occurring process which is defined as the degradation or
deterioration of a substance and or its properties usually a
metal over a period of time due to environmental exposure 1
this is an exergonic process as the metal tends toward the
lowest possible energy state

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