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 carboxylic acid phenol or sulfonic acid the other unknown in
 each sample is a neutral organic compound e g ketone ester
 amide etc the unknowns will be separated by extraction
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 phenol using the following ¹h nmr data is the unknown an
 alcohol or a phenol briefly explain your answer observed
 resonances doublet at 1 3 ppm septet at 2 8 ppm doublet at 3

7 ppm and a singlet at 4 1 ppm 2 you have an unknown
 organic compound that is either a ketone or an
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web the chromatography for gc and gc ms is identical in
 theory however the column used in the gc ms experiment is a
 capillary column as opposed to the packed column used in
 the gc experiment done in chemistry 105 a capillary column
 is simply a long tube made of glass with a small internal
 diameter for this experiment a 30 cm column with an

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 the reaction mechanisms an equally important part of
 organic chemistry has to do with the characterization and
 identification of compounds which may be encountered in
 sources ranging from a laboratory reaction to exotic tropical
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 organic compounds course chemistry institution bicol
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 identify the functional groups present in the given unknown
 and eventually identify what organic compound the unknown
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web identification of unknowns marks a point of mastery
 within an organic chemistry lab because it involves the
 unification of a series of systematic tests observations and
 analysis that only when considered all together can provide
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 points are assigned 5 pts total score 10 pts the liquid to its
 original bottle do not put other compounds or contaminated
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 relevant peaks labeled things listed above it was determined
 that the unknown molecule was piperonylamine and when
 looking up the molecule the molecule also had a melting
 point of 138 139 c and has a ama of 151

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 unknown organic compound purpose the purpose of this lab
 is to collect qualitative data on an unknown compound in
 order to discover the identity of the unknown via
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 important proficiency for any student completing the organic
 chemistry laboratory course this experiment culminates the
 laboratory techniques that you have learned this semester
 and requires you to put the various pieces of information
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web ms the spectrum for the unknown shows the molecular ion at 135 99 m z the base peak is at 56 98 m z which is the peak identified to be a butane chain of carbons the m 2 peak is at 138 99 m z and is roughly the same size as the m peak indicating the presence of bromine in this compound

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