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Daniela Niemeyer

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both compound types acids ionic

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can form compounds with different

ratios of their elements so prefixes are

used to specify the numbers of atoms of

each element in a molecule of the

compound examples include sf 6 sulfur

hexafluoride and n 2 o 4 dinitrogen

tetroxide

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containing monatomic ions consists of

the name of the cation the name of the

metal followed by the name of the anion

the name of the nonmetallic element

with its ending replaced by the suffix

ide some examples are given in table 2

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2 6 compounds containing

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mole of the compound would result in

two moles of bismuth a total of 417 96g

tellurium has a molar mass of one mole

of the compound would result in three

moles of tellurium a total of 382 8g add

the mass of bismuth and the mass of

tellurium per mole to find the total

molar mass divide the mass of bismuth

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