

# Hidroxido De Platino Iv

Thomas Griffiths

**25 Ejemplos de Hidróxidos** Ejemplos de hidróxidos Hidróxido de sodio, NaOH. Más conocido como soda cáustica, este hidróxido se usa para fabricar crayones, jabones, tinturas, explosivos, productos de petróleo y papel. También se usa en distintos tipos de limpiadores, detergentes y para destapar cañerías. El hidróxido de sodio es también usado en la industria textil, en la producción de diversos ... **Hidróxidos | Qué son, características, clasificación, cómo se ...** Los hidróxidos son un tipo de compuesto químico que está formado a partir de la unión de un elemento de tipo metálico o catiónico con un elemento que pertenece al grupo de los hidróxidos, o aniones. ¿Qué es el hidróxido? Hidróxido es un anión poliatómico con la fórmula química OH<sup>-</sup>. Está formado por un átomo de oxígeno y un átomo de hidrógeno unidos por un enlace covalente único. El hidróxido es la base conjugada del agua (H<sub>2</sub>O) y se forma cuando el agua pierde un ion hidrógeno (H<sup>+</sup>). El hidróxido es una base fuerte y un poderoso agente oxidante. Es corrosivo para los metales y puede ... Hidróxido: Definición y conceptos sobre hidróxidos Descubre todo sobre el hidróxido: definición, propiedades y conceptos clave de los hidróxidos. ¡Aprende y sorpréndete! **Hydroxide - Wikipedia** Hydroxide is a diatomic anion with chemical formula OH<sup>-</sup>. It consists of an oxygen and hydrogen atom held together by a single covalent bond, and carries a negative electric charge. It is an important but usually minor constituent of water. It functions as a base, a ligand, a nucleophile, and a catalyst. The hydroxide ion forms salts, some of which dissociate in aqueous solution, ... *Hidróxidos - Química Inorgânica - InfoEscola* Artigo sobre os hidróxidos, como são esses compostos químicos, características principais, como se formam, nomenclatura, entre outras informações. **Hidróxidos: qué son, propiedades, nomenclatura y ejemplos** Jan 14, 2022 · Los hidróxidos son compuestos inorgánicos y ternarios que consisten en la interacción entre un catión metálico y el grupo funcional OH (anión hidróxido, OH<sup>-</sup>). La mayoría de ellos son de naturaleza iónica, aunque también pueden tener enlaces covalentes.

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