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isotopes that have missing information fill in the missing information to complete the notation ${}^{143}_{54}\text{Xe}$ ${}^{14}_{54}\text{Xe}$ ${}^{36}_{16}\text{S}$ ${}^{36}_{16}\text{Mn}$ ${}^{57}_{26}\text{Fe}$ ${}^{57}_{26}\text{Mn}$ ${}^{121}_{50}\text{Sn}$ ${}^{121}_{50}\text{Sb}$ for each of the isotopes in question 21 2 3 determine the numbers of protons neutrons and electrons in a neutral atom of
nuclear chemistry vocabulary flashcards quizlet - Apr 11 2023
web nuclear chemistry vocabulary term 1 13 alpha particle click the card to flip definition 1 13 a positively charged atom that is released in the disintegration of radioactive elements and that consists of two protons and