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reduction is gain of e a oxidation b reduction c oxidizing
agent d reducing agent 2 write half reactions for each of the
following atoms or ions label each as oxidation or reduction
use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2
g h

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atoms change if an atom is oxidized its oxidation number
increases ie it becomes more ve or less ve if an atom is
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or more ve

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oxidation state of nitrogen in the following a no 3 b no 2 c no
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one or more atoms change oxidation numbers oxidation

occurs when an atom s oxidation state becomes more
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reactions are reactions that involve the transfer of electrons
from one species to another the species that loses electrons
is said to be oxidized while the species that gains electrons is
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recall that oxidation involves loss of electrons and reduction
involves gain of electrons represent simple redox reactions
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