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**Wörterbuch zur Erklärung und Verdeutschung der unserer Sprache aufgedrungenen fremden Ausdrücke ...**  
**Neue stark. verm. und verb. Ausg** Joachim-Heinrich von Campe,1813

*Der neu-testamentische Elias ...* Franz X. Hiltmayr,1738

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*Philothea* ,1852

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**Biblia, das ist die gantze Heilige Schrifft** Luther, 1708

**Parallel'nye slovari ĭazykov russkago, frantsuzskago, nĭmetskago i angliĭskago** Ch. Ph Reiff, 1853

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that no heat is lost to the surroundings

use the information below normal  
boiling point of nh 3 33 3 c  $\Delta h_{\text{vap}}$  23 4  
kj mol specific heat nh 3 l 4 74 j g k  
specific heat nh 3 g 2 06 j g k  
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web apr 3 2023 the experiment  
described above can be summarized in  
a graph called a heating curve figure  
below figure 11 16 1 11 16 1 in the  
heating curve of water the temperature  
is shown as heat is continually added  
changes of state occur during plateaus  
because the temperature is constant  
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imagine that you have a block of ice  
that is at a temperature of 30 c well  
below its melting point the ice is in a  
closed container as heat is steadily  
added to the ice block the water  
molecules will begin to vibrate faster  
and faster as they absorb kinetic  
energy

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explained and broken up into each  
component of phases and phase  
changes explaining what equations to  
use what specific heat is

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web the cooling curve a plot of  
temperature versus cooling time in  
figure pageindex 4 plots temperature  
versus time as a 75 g sample of steam  
initially at 1 atm and 200 c is cooled  
although we might expect the cooling  
curve to be the mirror image of the  
heating curve in figure pageindex 3 the  
cooling curve is not an

## **8 10 cooling curves chemistry**

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web once all of the material has  
solidified at the time indicated by point

c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase

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web heating and cooling curves are graphs they plot a substance s temperature y axis against heat x axis for heating curves we start with a solid and add heat energy for cooling curves we start with the gas phase and remove heat energy cooling and heating curves have five segments let s study each one by looking at heating ice to

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web for water this temperature is 100 c because the boiling point for water is

100 c different substances have different melting points and boiling points but the shapes of their heating curves are very similar for example this is the heating curve for iron a metal that melts at 1538 c and boils at 2861 c cooling curves

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