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 but different masses due to different
 numbers of neutrons in the nuclei the
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 different isotopes

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 most people respond that you add up
 the values and divide by how many
 there were the average of 2 2 4 and 8 is
 16 4 or

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web procedure 1 obtain a sample of
 candium 2 separate the three isotopes
 and count obtain the total mass for
 each group of isotopes 3 complete the
 data table using the instructions found
 under the data analysis questions
 section the large x s in the table
 indicate cells that do not require data
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web candium is an interesting element
 found only at fordham high school for
 the arts we will be using this special
 new element today to learn about
 atomic mass calculations the atomic
 mass listed on the periodic table for
 each element is a weighted average
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web 1 obtain a sample of candium
 separate the 3 isotopes candium red
 hots mini candium peanut butter
 regular and candium peanut large and
 record the numbers for each type of
 isotope 2 next determine the mass for
 each isotope and calculate the total
 mass in grams 3 calculate the average
 mass

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web obtain a sample of candium from your teacher 2 separate the three isotopes m m s skittle s and peanut m m s and measure the mass of each isotope record in data table 1 1 3 count the number of m m s skittle s and peanut m m s record in data table 1 1

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web procedure 1 place the candium atoms in a paper towel or a small bag 2 gently shake the atoms for 30 seconds this is the estimated half life of your candy 3 gently pour out the atoms onto

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