

Ley De Conservacion De Masas

Robert W. Parry

Quimica,2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Fundamentos de Quimica Ralph A. Burns,2003

Principios básicos de química Harry B. Gray,Gilbert P. Haight,1975 Este texto es una respuesta al reto que significa presentar una relación innovadora y moderna de los principios de la Química. El rápido desarrollo de los últimos cursos preuniversitarios está creando presiones sin precedentes que obligan a cambiar la enseñanza en el curso de iniciación. En este nivel se necesitan urgentemente aproximaciones al estudio de los campos más avanzados de la Química, con una presentación puesta al dia. Además del texto propiamente dicho, existe la Guía para el profesor.

Termodinamica Kurt C. Rolle,2006-09-08 Este conocido texto, diseñado especialmente para programas de ingeniería y tecnología, presenta los conceptos fundamentales de termodinámica y sus aplicaciones prácticas a la energía térmica, a la transferencia de calor y a la calefacción y acondicionamiento de aire. Contiene problemas de ingeniería y diseño en el mundo real, y no abstractas matemáticas.

Introducción a los conceptos y teorías de las ciencias físicas Gerald James Holton,Stephen G. Brush,1996-10 Quizás la característica más sorprendente de este libro es el énfasis que pone en la naturaleza del descubrimiento, el razonamiento y la formación de conceptos como un tópico fascinante. Esto significa que los aspectos históricos y filosóficos de la exposición no son meramente un ingrediente dulzón para conseguir que el lector digiera el texto lo más fácilmente posible, sino que se presentan por su propio interés intrínseco.

Química G. William Daub,William S. Seese,1996

Quimica: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and

Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Cálculo de balances de materia y energía Ernest J. Henley, 1973 Este texto corresponde a un curso de Estequiométría y Termodinámica. Se supone que los estudiantes han de estar capacitados para saber cómo los balances de materia y energía se transforman en componentes integrales de los sistemas de diseño con ayuda de máquinas calculadoras, que están asumiendo una parte importante y creciente de nuestra experiencia profesional. Sin embargo, se ha procurado no perder de vista los métodos tradicionales.

Fisicoquímica Gilbert W. Castellan, 1998

Química General Práctica,

Mecánica newtoniana A. P. French, 1974 Consultar comentario general en la obra completa.

Química. Fundamentos experimentales Robert W. Parry, 1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseanza donde se cursan estudios de nivel medio y preuniversitario. Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día. Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

Química. Fundamentos experimentales. Guía del profesor Robert W. Parry, 1974 Esta Guía está destinada a proporcionar apoyo al profesor y a establecer claramente los principios y las razones que guiaron a los autores en la programación y desarrollo del libro Química, fundamentos experimentales.

Química General Manual de Prácticas de Laboratorio Instituto Tecnológico de Santo Domingo,

Química para el nuevo milenio John W. Hill, 1999

Operaciones básicas de ingeniería química Warren L. McCabe, Julian C. Smith, 1981 En esta obra se ha proseguido la tradición de libros más antiguos sobre esta materia, ya que, durante la pasada generación, las operaciones básicas han demostrado su utilidad tanto en la enseñanza como en la práctica de la ingeniería. Si bien algunas operaciones especialmente absorción de gases, destilación y extracción muestran tendencia a fusionarse, hemos creído conveniente conservar la

integridad de las operaciones individuales y estudiarlas separadamente, ya que cada una de ellas se realiza en la práctica de una forma peculiar.

Principios de los procesos químicos .balances de materia y energia, VOL 1 Olaf A. Hougen,R. A. Ragatz,1982 Consultar comentario general de la obra completa.

Química 1 Cuevas Quintero Antonio,2003

Química: teoría y problemas José María Teijón,José Antonio García Pérez,1996 Explica cada uno de los conceptos de la Química General y los aplica en más de 500 problemas resueltos. Incluye prácticas de laboratorio y tests de autoevaluación. También adecuado para COU-LOGSE.

Aprender y enseñar ciencia Juan Ignacio Pozo Municio,Juan Ignacio Pozo,Miguel Ángel Gómez Crespo,1998 En una sociedad en donde la ciencia está cada vez más presente en la vida cotidiana, parece necesario hacerla accesible a la mayor parte de los alumnos. No obstante, muchos datos revelan que éstos, en vez de asimilar las teorías y modelos científicos enseñados en clase, siguen interpretando el mundo según esquemas intuitivos o culturales ajenos a la ciencia. Por otra parte, cada vez hay más profesores de ciencias que comprueban, con frustración, que sus estudiantes apenas están interesados en ese saber científico y tienen también serias dificultades para utilizarlo en la resolución de problemas escolares o cotidianos. La respuesta a esta crisis de la educación científica debe basarse en su renovación profunda que, sin renunciar a sus propósitos y contenidos esenciales -hacer participar a los alumnos y futuros ciudadanos de las actitudes, las formas de pensamiento y los modelos y teorías propios de la ciencia como discurso social- se apoye en la psicología de los alumnos. Este libro, producto de la cooperación de un psicólogo con un profesor de química, propone un acercamiento entre la ciencia y la mente de los estudiantes, necesario para los profesores de ciencias, ya que suelen tener una formación específica en su materia pero un menor conocimiento de la psicología y de la didáctica que les ayude a beneficiarse de ese saber disciplinar. No se trata de proporcionar recetas ni soluciones preconcebidas, sino de ayudar a esos docentes a dar más sentido a su práctica y a superar las dificultades que conlleva, comprendiendo cómo aprenden ciencia sus alumnos y cómo, a través de su enseñanza, pueden ayudarles a asimilarla mejor.

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