

Oh The Places Youll Go Quotes

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What are the spectator ions in the reaction between sodium Oct 16, 2016 · The nitrate and the natrium ions. $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$ The net ionic equation is: $2\text{Ag}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Ag}_2\text{CO}_3(\text{s})$... **What is the product of the following reaction? 1)CH₃ OH - Socratic** Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $\text{NH}_4^+ + \text{HO}^- \rightarrow \text{NH}_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$ For methanol, the acid base reaction would proceed ... **Question #18488 - Socratic** The degree of dissociation α (alpha=0.0158) α (K_b=2.51x10⁻⁶) color (white) (x)"mol/l") Triethylamine is a weak base and ionises: $(\text{CH}_3)_3\text{N} + \text{H}_2\text{O} \rightleftharpoons (\text{CH}_3)_3\text{NH}^+ + \text{OH}^-$... Question #627ef - Socratic functional groups like -OH, -CHO, -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO, or -OH and -CO-. Aldose shows different reactions ... Question #22e51 - Socratic Products formed are; Cu(OH)₂ + HCl → CuCl₂ + H₂O Since CuCl₂ is soluble in water.. But; Cu(OH)₂ + HCl → CuCl₂ + H₂O " [Not Balanced]" Balancing the equation.. *In each of the following pairs, which one has the lower viscosity?* Jul 31, 2015 · C₂H₅Cl has weak dipole-dipole forces from the C-Cl bonds, but CH₃-OH has strong hydrogen bonds. Confirmation: The viscosity of C₂H₅Cl is 0.279 mPa·s, while the viscosity of ... **Could a buffered solution be made by mixing aqueous ... - Socratic** Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ... **Question #44ec4 - Socratic** That's a silly way to say carbon dioxide! $\text{CO}_2(\text{g}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq})$ May be wrong, but it seems reasonable. The gas would interact ... Question #71ce2 - Socratic Nov 9, 2014 · $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$ when the acid was added to the resulting solution. The H^+ and OH^- react in a 1:1 ratio. This tells us that the number of moles of H^+ used will be equal ... **What is the chemical formula for lithium hydroxide? - Socratic** Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M⁺ ion. Hydroxide anion, OH⁻, has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ...

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