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benzaldehyde and acetone under basic conditions the gc ms
was used to relate the molar mass of the product thin layer
chromatography was used to visualize the relative rf factor
for each isomer uv vis instrument showed three peaks that
were assigned to each isomer

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and benzaldehyde to make dibenzalacetone the percent yield
of dibenzalacetone was found to be 73 which is a good
percent yield the melting point of dibenzalacetone was
around the range of 104 110 c this helped conclude that
dibenzalacetone had a trans trans e e isomer

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along with the loss of water for example in the aldol
condensation shown in figure 1 two molecules of ethanal are
reacted to produce an a b unsaturated aldehyde and water
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