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Julio De Paula

Introduccion a la Fisicoquimica: Termodinamica Thomas Engel,Philip J. Reid,2007

Bioquímica humana. Curso básico. José M. Macarulla,Félix M. Goñi,1994-06 El libro está dirigido sobre todo a profesionales sanitarios; las Escuelas de Enfermería lo encontrarán especialmente adecuado para sus cursos de Bioquímica y también será de utilidad para estudiantes de Medicina, e incluso médicos en ejercicio que deseen actualizar conocimientos. El diseño del libro permite estudiarlo a distintos niveles de profundidad. Las figuras añaden a menudo detalles no incluidos en el texto básico, y lo mismo ocurre con los problemas y prácticas de laboratorio. Además, hay que tener en cuenta las secciones de complementos y aplicaciones clínicas.

Bioquímica Cuantitativa José M. Macarulla,Aída Marino,Alberto Macarulla,1992 Consultar comentario general de la obra completa.

Química-física practica de findlay Alexander Findlay,B. P. Levitt,1979 La Química física práctica de Findlay habrá de ser una guía esencial para los estudiantes de Química. Esta edición española corresponde a la novena inglesa y en ella se ha ampliado su contenido hasta el nivel en el que la mayor parte de las técnicas que se utilicen, se puedan ofrecer y realizar en el tiempo normalmente asignado a la Química práctica.

Quimica: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry.For anyone interested in learning more about chemistry.

Atkins química física Julio De Paula,2000 Química Física de Atkins continúa siendo el estándar a emular en el contexto de

un curso de Química en todo el mundo. La elección atinada de los temas, el estilo de redacción claro de los autores y la exposición minuciosa de las matemáticas reafirman la posición del libro como un líder del mercado.

Diccionario de ciencias,2000

Fisicoquimica Gilbert W. Castellan,1998

Sistemas Químicos José Masaguer Fernández,1966

Problemas de química Henry O. Daley,Robert F. O'Malley,1979 El objeto del presente libro, no sólo es facilitar al estudiante de los primeros cursos un número más o menos grande de problemas que ilustren los principios teóricos de la Química sino también enseñarle a resolver estos problemas.

Fisiología animal comparada Richard W. Hill,2021-03-01 Este libro es una introducción básica para aquellos aspectos de la fisiología animal comparada que se apoyan particularmente en las interrelaciones fisiológicas entre los animales y su ambiente. Los atributos fisiológicos de los animales son discutidos en vistas a su significación en el teatro ecológico y se ha hecho énfasis en la perspectiva evolucionista ya que es la interacción histórica entre el animal y el ambiente la que ha condicionado la forma y la Fisiología de las especies modernas.

Principios de los procesos químicos .balances de materia y energia, VOL 1 Olaf A. Hougen,R. A. Ragatz,1982 Consultar comentario general de la obra completa.

Química. La ciencia básica DOMINGUEZ REBOIRAS, MIGUEL ANGEL,2006-01-01 Esta obra se ha escrito pensando en las necesidades de los estudiantes de los primeros cursos de Facultades de Ciencias, Farmacia, Ciencias Ambientales y de la Salud, y Escuelas Técnicas.

Técnicas y métodos de investigación en nutrición humana Alberto Miján de la Torre,2002 Contiene: Técnicas de investigación en nutrición humana: Nutrición y genética: nutrigenómica y técnicas asociadas; Bioquímica en nutrición: minerales y oligoelementos; Bioquímica en nutrición: vitaminas; Cultivos celulares en nutrición; El animalario en nutrición aplicada; El cuerpo humano: técnicas de estudio de la composición corporal; El cuerpo humano: técnicas de estudio del consumo energético; El cuerpo humano: metabolismo proteico; El cuerpo humano: estudio del tejido adiposo; El cuerpo humano: sistema óseo; Evaluación de la función digestiva.- Métodos de investigación en nutrición humana: Guía de los objetivos en nutrición clínica. Conceptos de causalidad y standar de oro; Conceptos estadísticos; Estadística aplicada en el diseño y la ejecución de estudios nutricionales; Epidemiología nutricional. Estudios descriptivos; Estudios observacionales. Estudios de cohortes y estudios de caos y controles; Diseño experimental. El ensayo clínico aleatorizado; Metaanálisis y análisis multivariante...

Química e investigación criminal Matthew E. Johll,2008 La fascinación de la sociedad por las investigaciones criminales se refleja en la popularidad de las novelas y series de televisión de gran audiencia basadas en la resolución de crímenes y el

trabajo policial. Las historias de Sherlock Colmes, las novelas de misterio de Agatha Christie, las aventuras de investigadores inolvidables como Nancy Drew, Dick Tracy, Perry Mason y Colombo, y últimamente, la exitosa serie de televisión CSI: Crime Scene Investigation, que intriga a millones de personas, son claros ejemplos. El profesor Matthew E. Johll ha encontrado en este interés por la ciencia forense una excelente oportunidad para escribir su libro Química e investigación criminal con la finalidad de atraer a los estudiantes hacia el estudio de la química. Así, basándose en casos reales de la ciencia forense y la investigación criminal, explica los principios químicos básicos de una manera clara y accesible, sin olvidar los aspectos más cuantitativos de la química.

Centrales de vapor G. A. Gaffert, 1981

Fundamentos de la química general e inorgánica Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos- ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dicho- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

Química general M. Garric, 1979 La novedad más importante de este curso de Química general con respecto a otros similares consiste en la introducción de los conocimientos más modernos en temas tales como: 1. Estereoquímica, estructuras iónicas y covalentes, ilustradas por más de 120 figuras. 2. Teorías del enlace químico expuestas mediante algunos recursos simples de Mecánica cuántica y con numerosos ejemplos clásicos. 3. Estudio de las soluciones acuosas siguiendo a Brönsted empleando los métodos gráficos de la escuela escandinava. 4. Titulaciones ácido-base, titulaciones por precipitación, complejométricas y redox, desarrolladas especialmente en los textos norteamericanos.

Química: libro del alumno Nuffield Foundation, 1984 Consultar comentario general de la obra completa.

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