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AW Chickering

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remains of the shell you may have to rub a little gently 4  
carefully use the tape measure to check the circumference of  
the egg record this data step 2 osmosis experiment 1

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beaker of acid 1 5 m 2 m of hydrochloric acid is ideal this will  
dissolve the shells weigh down the eggs with a second  
beaker containing water so they are pushed down into the  
acid and do not float above the surface this ensures all the  
shell is removed and there is not a patch left

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eggs effect of solute concentration upon rate degree of  
osmosis in chicken eggs introduction if a cell is to perform its  
functions it must maintain a steady state in the midst of an  
everchanging environment

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in chicken eggs all students will place two eggs in water  
each student will also test at least one other solution of their  
own choosing on two other eggs you may even choose

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widest point using a tape measure or a piece of string to use

a piece of string take a piece of string and wrap it around the  
egg once mark with your finger the length of string that is  
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into the base of the container make sure that the egg does  
not crack if it does replace the egg step 2 pour vinegar on  
top of the egg making sure that the egg is fully submerged  
step 3 leave the egg in the vinegar for about 24 hours step 4  
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create our very own model of a cell using a shell less chicken  
egg we can see what happens when we manipulate the  
osmotic balance in the cell materials 3 eggs 3 glasses large  
enough to fit the egg plus liquid 3 butter knives white

vinegar about 3 cups distilled water about

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web weigh your egg and note the measurement put your naked egg in a jar and add enough corn syrup to cover the egg store the egg in a refrigerator or somewhere cool for 24 hours after 24 hours scoop out the egg and observe the changes weigh the egg again and note the measurement extensions draw a diagram of your egg in the corn syrup

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