

Ejercicios Leyes De Gases

Christopher J. Willis

Problemas de química adaptados al curso de principios de química Ian S. Butler, Arthur E. Grosser, 1976 Contiene: Átomos, moléculas y moles; Las leyes de los gases y la teoría cinética; Materia con carga; Cantidades en reacciones químicas: estequiométría; Equilibrio químico; Clasificación de los elementos y propiedades periódicas; Reacciones redox; Teoría cuántica y estructura atómica; Estructura electrónica y propiedades químicas; Enlace covalente; Compuestos de coordinación; La misión especial del carbono; Química nuclear; Enlaces en sólidos y líquidos; Termodinámica; Energía libre y equilibrio; Equilibrios de oxidación-reducción y electroquímica; Cinética química; Problemas de examen; Cálculos matemáticos.

Resolución de problemas de química general Christopher J. Willis, 1980 Este libro acerca de la resolución de problemas intenta ser un complemento de los textos elementales de Química. Incluye la mayoría de los temas que se consideran durante un curso, y resulta aconsejable tanto para estudiar bajo la dirección del profesor, como para hacerlo el alumno por su propia cuenta o con un mínimo de supervisión.

Resolución de problemas de química general R. Nelson Smith, Conway Pierce, 1991 Este libro será útil a aquellos alumnos que empiezan a estudiar Química y tienen dificultades al analizar los problemas y encontrar soluciones lógicas o que tienen dificultades con las representaciones gráficas y su interpretación.

Química G. William Daub, William S. Seese, 1996

Química: teoría y problemas José María Teijón, José Antonio García Pérez, 1996 Explica cada uno de los conceptos de la Química General y los aplica en más de 500 problemas resueltos. Incluye prácticas de laboratorio y tests de autoevaluación. También adecuado para COU-LOGSE.

Química, 2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

Problemas de química Henry O. Daley, Robert F. O'Malley, 1979 El objeto del presente libro, no sólo es facilitar al

estudiante de los primeros cursos un número más o menos grande de problemas que ilustren los principios teóricos de la Química sino también enseñarle a resolver estos problemas.

Termodinámica: 100 ejercicios y problemas H. Lumbroso, 2021-01-10 Esta obra de Termodinámica va destinada a los estudiantes de los cursos preparatorios de las Escuelas de Ingenieros, así como a los del primer ciclo de las Facultades de Ciencias.

Physics Douglas C. Giancoli, 2005 Presents basic concepts in physics, covering topics such as kinematics, Newton's laws of motion, gravitation, fluids, sound, heat, thermodynamics, magnetism, nuclear physics, and more, examples, practice questions and problems.

Fundamentos de Química Ralph A. Burns, 2003

Física preuniversitaria. I Paul A. Tipler, 1991 Este libro se ha escrito para estudiantes que cursan la asignatura de Física en la enseñanza preuniversitaria. En él se desarrolla la física elemental necesaria para futuros estudiantes de Biología, Medicina, Física, Química, Ingeniería, etc... Los te

Problemas de química Michell J. Sienko, 1971 Durante la última década, y particularmente en los Estados Unidos, ha surgido un magnífico autoperfeccionamiento dentro de la enseñanza de la Química elemental. Sin embargo, a medida que ha aumentado el nivel de complicación, mostrando ya desde el primer curso, y cada vez más, los sútiles y profundos aspectos de la Química, ha hecho aparición una tendencia creciente a dejar de lado los aspectos primitivos del cálculo químico.

Problemas de Química Física, 2006 Esta obra que usted tiene en sus manos responde a la necesidad detectada por los autores de llegar claramente al alumno de Química Física con un manual práctico que responda a la propia naturaleza de la disciplina y le permita su correcta comprensión entendiendo, de manera precisa y completa, cómo se solucionan los problemas planteados en el aula o en el laboratorio. De ahí, este extraordinario esfuerzo por llevar al estudiante, fundamentalmente, una selección de los mejores problemas de imprescindible resolución si se quiere garantizar la correcta preparación de aquel que quiera continuar y profundizar en el apasionante mundo de la Química, desde las diferentes perspectivas, en ciencias puras y/o aplicadas. Tomado de: <http://bit.ly/2gZiZHU>.

Problemas Fundamentales en Meteorología,

115 ejercicios resueltos de programación C++ Irma Yolanda Polanco Guzman, Jorge Fernando Betancourt Uscategui, 2022-01-28 El presente libro está orientado a brindar los conocimientos del lenguaje de programación C++ versión 2019. A través de ejercicios resueltos aplicados a diferentes temáticas, se aborda cada uno de los contenidos desde lo básico hasta la programación orientada a objetos POO. En el capítulo 1 se introduce al lector en lenguaje de programación C++, describiendo las principales características y generalidades del lenguaje y a partir del capítulo 2 se presenta ejercicios resueltos relacionados con: Capítulo 2. Estructuras secuenciales Capítulo 3. Estructuras selectivas if, if - else e if anidados

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Quimica Para El Acceso a Ciclos Formativos de Grado Superior .e-book. María Dolores Santana,2004-05-20

Quimica: la Ciencia Central Theodore L. Brown,H. Eugene LeMay,Bruce Edward Bursten,Julia R. Burdge,2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish.Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry.For anyone interested in learning more about chemistry.

El desarrollo de habilidades para la resolución de problemas en la Ingeniería Química Armando Rugarcía Torres,1993

La Química en problemas José María Teijón,2006 Este libro complementa los conocimientos teóricos de los estudiantes de Química General de primeros cursos de facultades de Ciencias, Escuelas Universitarias y Escuelas Técnicas Superiores, y resuelve a lo largo de sus 480 páginas muchos de los problemas que han sido propuestos en exámenes realizados en estos centros universitarios.

Atkins química física Julio De Paula,2000 Química Física de Atkins continúa siendo el estándar a emular en el contexto de un curso de Química en todo el mundo. La elección atinada de los temas, el estilo de redacción claro de los autores y la exposición minuciosa de las matemáticas reafirman la posición del libro como un líder del mercado.

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