

# Kokain Reinheit Testen Zuhause

**Kathleen Armour**

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the final temperature is exactly halfway  
between  $t_1$  and  $t_2$  the final  
temperature is closer to  $t_2$  than to  $t_1$  c  
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web in this set of practice questions we  
will go over the main types of questions  
on calorimetry including the heat  
capacity the heat of reaction finding the  
final temperature of a mixture constant  
pressure calorimetry and constant  
volume calorimetry a few important  
concepts and formulas you will need to  
solve these questions  
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moles of hexane  $C_6H_{14}$  is combusted in  
a bomb calorimeter containing 5.65  
liters of water calculate the molar heat  
of combustion of hexane if the water  
temperature rises 5.54 °C the specific  
heat capacity of water is 4.184 J/g °C  
ms t h 5.650 grams  $H_2O$  4.184 J/g °C 5.54  
°C h 1310 kJ  
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web 6 a 218 g sample of steam at 121°C  
is cooled to ice at 14°C find the change  
in heat content of the system 7 in going  
from ice at 34°C to steam at 138°C a  
sample of water absorbs  $1.41 \times 10^5$  J  
find the mass of the sample answers 1 2  
68 x 104 J 2 6 23 x 104 J 3 1 71 x 104 J 4  
1 11 x 105 J 5 2 11 x 105 J 6  
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