

Kidnapped Robert L Stevenson

Marco Cascella

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p4o10 pf5 scl2 sulfur dichloride n2o4 dinitrogen tetroxide

the a of tetra is dropped before the o of oxide p4o10

tetraphosphorous decoxide the a of deca is dropped before the o of oxide pf5 phosphorous pentafluoride

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web acids are covalently bonded molecules but when they are put into water they produce ions one of the ions produced is always h which immediately combines with a water molecule to form the hydronium ion H_3O^+ the H_3O^+ ion is what defines the acidic properties of a substance

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web examine the pairs of ternary acids in model 2 that contain sulfur phosphorus and nitrogen each pair has one acid that ends in ic and another that ends in ous these endings are related to the name of the polyatomic anion found in the acid ate or ite

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web the acids lose a hydrogen ion for each acid base reaction in model 2 describe the role of the bronsted lowry base in the proton H^+ ion transfer that occurs the base gains or accepts a hydrogen ion proton as you saw in last week's pogil all arrhenius bases in model 1 have an OH^- ion in their chemical formulas

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