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**Judd E. Hollander**

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we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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web feb 14 2014 we have therefore started our studies by identifying a theory level that properly describes the competition between the pathways a d presented in fig 1 fig 1 possible reactions of toluene and with permanganate anion at positions a c m b c 1 c 2 c c 2 c 3 d c 3 c 4 full size image

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observed

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web the structure of an organic compound which on oxidation gives an acid that gives a single mono substituted product on nitration with hno<sub>3</sub> and h<sub>2</sub>so<sub>4</sub> is medium view solution

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web may 15 2020 toluene 1 oxidation with kmno<sub>4</sub> mechanism 2 benzoic acid c<sub>6</sub>h<sub>5</sub>cooh socl<sub>2</sub> 3 benzoyl chloride c<sub>6</sub>h<sub>5</sub>cocl lialh oc ch<sub>3</sub> 3 h 4 benzaldehyde c<sub>6</sub>h<sub>5</sub>cho ch<sub>2</sub>ohch<sub>2</sub>oh h step 1 toluene oxidation with kmno<sub>4</sub> kmno<sub>4</sub> is a strong oxidising agent

which oxidise toluene to benzaldehyde in 1st step

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