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Ying-Ying Zheng

What is the chemical formula for lithium hydroxide? - Socratic Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M^+ ion. Hydroxide anion, $-OH$, has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ... What reaction represents formation of $Cu(OH)_2$? | Socratic Copper forms an insoluble hydroxide that is fairly poorly characterized; another way we could look at it as a hydrated copper oxide, i.e. $CuO \cdot OH_2 \equiv Cu(OH)_2$. *Calculating the concentration of excess HCL in E - Socratic* The acid in excess is then titrated with $N aOH(aq)$ of KNOWN concentration....we can thus get back to the concentration or molar quantity of $M(OH)_2$...as it stands the question (and answer) ... **Question #d6b18 - Socratic** We want the standard enthalpy of formation for $Ca(OH)_2$. Thus, our required equation is the equation where all the constituent elements combine to form the compound, i.e.: $Ca \dots$ What is the product of the following reaction? 1) $CH_3OH - Socratic$ Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $NH_4^+ + HO^- \rightleftharpoons NH_3(aq) + H_2O(l)$ For methanol, the acid base reaction would proceed ... *$C_3H_7OH + O_2$ synthesis? - Socratic* Apr 19, 2018 · See below When a substance rapidly reacts with oxygen gas, an alcohol in this case, we call it combustion, so that will be your reaction type and not synthesis When ... What is this oxidation number of CH_3OH and CH_2O ? | Socratic Jun 29, 2016 · Given that oxidation number is an empirical concept; the oxidation number of C in methanol is -II; in formaldehyde it is 0. Oxidation number/oxidation state is an empirical ... **How do acid and bases react? + Example - Socratic** Mar 22, 2016 · When an acid and a base are placed together, they react to neutralize the acid and base properties, producing a salt (neutralisation). The H^+ in the acid combines with the OH^- (... *Which is the stronger acid in each of the following pair* Jul 31, 2015 · For oxyacids with the same number of OH groups and the same number of O atoms, acid strength increases with increasing electronegativity of the central atom. Since P is ... **What are the spectator ions in the reaction between sodium** Oct 16, 2016 · The nitrate and the sodium ions. $Na_2CO_3(aq) + 2AgNO_3(aq) \rightleftharpoons Ag_2CO_3(s) + 2NaNO_3(aq)$ The net ionic equation is: $2Ag^+(aq) + CO_3^{2-}(aq) \rightleftharpoons$...

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