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equation for a reaction when you see a chemical equation in a textbook it looks like the reactants are completely used up and all made into product

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and products are typically measured by mass or by volume it is first necessary to convert the given quantities of each reactant to moles in order to identify the limiting reactant

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mgo is formed 4

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lower than required by the reaction
stoichiometry thus limiting the amount
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reactants and products are represented
as neutral substances

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calculation whichever reactant gives
the lesser amount of product is the
limiting reagent what we need to do is
determine an amount of one product
either moles or mass assuming all of
each reactant reacts

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obtained is called the limiting reactant
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for dinner and want to bake brownies

for dessert

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the key to recognizing which reactant is
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whichever reactant gives the lesser
amount of product is the limiting
reagent what we

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limiting reagent the amount of product
formed by each reactant is calculated
the limiting reactant is the reactant
from which the minimum amount of
product is formed also if we calculate
the amount of

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