

# Hyatt House Philadelphia Plymouth Meeting

**Annelies Wilder-Smith**

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web answer see below explanation the rate of a chemical reaction  $A + B \rightarrow C + D$  is given in first approximation by  $v = k[A]^m[B]^n$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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