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 $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$  molar mass of  $\text{KClO}_3$  122.5 g/mol  $\text{KCl}$  74.55 g/mol

02 31 998 g/mol if the mass of potassium chlorate was 240  
grams which of the following calculations can be used to  
determine the mass of oxygen gas formed

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potassium chloride is emptied into the water data will be  
collected for 180 seconds 5 using a glass stirring rod gently

stir the water in the beaker for about 20 seconds to help the potassium chloride dissolve 6 after 180 seconds have lapsed the calculator

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