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known that 1 mol N_2 2 mol NH_3 use the periodic table to look at the atomic weights of the elements and calculate the weights of the reactants and products 1 mol of N_2 28.0 g 2 mol of NH_3 is 34.0 g 1 mol of H_2O is 18.0 g

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web solution average atomic mass of Li 6.941 amu $\times 6.0151$ amu $\times 7.0160$ amu $\times 6.941$ $\times 1.0009$ $\times 7.0160$ $\times 1.0009$ $\times 0.075$ $\times 0.075$ corresponds to a natural abundance of 6Li of 7.5 percent the natural abundance of 7Li is $\times 0.925$ or 92.5 percent 3.7 the unit factor required is 6.022

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web a molar mass Li_2CO_3 73.89 g b molar mass CS_2 76.15 g c molar mass CHCl_3 119.37 g d molar mass $\text{C}_6\text{H}_8\text{O}_6$ 176.12 g e molar mass KNO_3 101.11 g

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