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conditions of p and t the particles in an ideal gas do not have finite size and volume the collisions between the ideal gas particles are said to be elastic they exert no attractive or repulsive forces hydrogen gas generated in today s experiment is however a real gas not an ideal gas *determining the molar mass of butane lab answers studocu* - Aug 16 2023

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web although the most convenient gas for this is butane other gases may be available in similar small cylinders the collection of a gas sample and the weighing of the gas cylinder before and after this should take about 5 10 minutes as a demonstration